NEW SESQUITERPENE LACTONES FROM INULA SPECIES*

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Key Word Index—*Inula* species; Compositae; new eudesmanolides; new germacranolides; new guaianolides; new pseudoguaianolide, new xanthanolides.

Abstract—An investigation of *Inula* helenium, *I. royleana*, *I. salicina* and *I. bifrons* afforded in addition to known sesquiterpene lactones 20 new lactones, the eudesmanolides 3, 6 and 8–12, the germacranolides 14–18 and 20–22, the guaianolides 23 and 25, the pseudoguaianolide 26, the xanthanolides 28 and 32 and the cyclopropane analogue 30. Structures and configurations of these new compounds were established by extensive PMR studies and by some chemical transformations. Some of the investigated species contain besides the widely distributed pentaynene 33 several known thymol derivatives together with a new one (40).

INTRODUCTION

The genus Inula (tribe Inuleae) with about 80–100 species represents a badly delineated complex [1]. Chemical investigations have shown that there are two main groups, one containing sesquiterpene lactones, especially eudesmanolides, and one containing simple thymol derivatives [2]. Furthermore several triterpenes and some diterpenes are present, while the pentaynene 33 seems to be widespread [2]. Reinvestigation of 3 species together with the investigation of a further one has now led to the isolation of a considerable number of new sesquiterpene lactones.

RESULTS AND DISCUSSION

Investigation of *Inula helenium* L., *I. royleana* DC, *I. salicina* and *I. bifrons* L., all grown from seeds, afforded together with large amounts of alantolactone (5) and isoalantolactone (1) many sesquiterpene lactones, most of them not previously isolated. Scheme 1 contains the lactones isolated and some transformation products and scheme 2 the other compounds. Table 8 lists the species investigated with the isolated compounds.

The fresh plant material was extracted with mixtures of ether and petrol; the extracts were first separated by column chromatography and further by TLC, which normally had to be repeated several times as very often extremely complex mixtures were obtained. In such instances, AgNO₃ columns and plates were frequently useful. Besides MS data 270 MHz-PMR spectra were extremely useful in structural elucidation of the new lactones. In most cases assignment of signals was possible by double resonance experiments and application of shift reagent.

The structure of 3 could be solved easily as the spectral data of ivalin acetate (2) [13] were identical with those of the isolated lactone (see Table 1). Structure and

*Part 142 in the series "Naturally Occurring Terpene-Derivatives"; for part 141 see: Bohlmann, F. and Zdero, C. Phytochemistry. 17, 759. configuration of 6 was elucidated by extensive decoupling experiments of the PMR spectra of both 6 and the corresponding ketone 8, which was not only prepared by chromic acid-pyridine complex oxidation, but also isolated as a natural product (see Table 1). The cisannelation of the lactone group of 6 follows from the value of $J_{7\alpha,8\alpha}$ and $J_{8\alpha,9}$. They are always smaller than those for $J_{7\alpha,8\beta}$ and $J_{8\beta,9\alpha}$ (see Tables 1 and 2). The β -orientation of the 4-methyl group follows from the couplings too. Only the equatorial position of the 2-hydroxy group is in agreement with the observed couplings $(J_{1\alpha,2} = J_{2\alpha,2} = 10)$.

couplings $(J_{1\alpha,2} = J_{2,3\alpha} = 10)$. Similar arguments for another compound led to structure 9, which is that of hydroxyalantolactone. The observed shifts after addition of Eu(fod)₃ are in good agreement with 9 (see Table 1).

The structure of 10 was elucidated by double resonance experiments which allowed assignment of most PMR signals. The 8\alpha-H signal is a ddd at 4.5 with equal coupling constants, which indicates cis-annelation of the lactone ring. Furthermore the chemical shift of 7α -H is typical of cis-fused lactones whereas in trans-lactones the signal of 7α -H is always shifted to considerably higher fields. The position of the double bond clearly follows from the 6-H signals (see Table 2). The enantiomer of 10 has been reported as constituent of liverworts [23]. All PMR data of 11 are in good agreement with the proposed structure. They are similar to those of 10, but due to the deshielding effect of the OH group the 9-H signals are now separated, which further establishes the β -orientation of the OH. 11 is the 8-epimer of ivangustin [3] which also follows from the chemical shift of 7α -H. 12 is an unsaturated analogue of 11. The PMR data (Table 2) clearly show that a trans-fused lactone is present; the β -position of the 1-OH follows from the downfield shift of the 9β -H.

While 2α -OH costunolide (19) is known [5], the 9β -derivatives 14-18 are new. The 9β -position of the oxygen function can be inferred from the shifts which result after addition of Eu(fod)₃ as both the 10-Me and 1-H signal, are shifted downfield. If the oxygen function

Scheme 1 (literature citations in brackets).

28: R = OH, R' = H **29**: R = R' = 0

30: R = OH, R' = H31: R = R' = O[15]

32

^{*} Prop = Co Et; Mebu = COCH(Me)Et; iVal = COCH₂CHMe₂; iBu = COCHMe₂.

Scheme 2.

were at C-3 the signal of the 4-Me group should be shifted (see Table 3). All other signals are in good agreement with the proposed structures.

Structure and configuration of 20 were confirmed by extensive decoupling experiments. All data are in good agreement with the proposed structure (see Table 4). Assignment of the signals was confirmed by using the solvent shift method. The observed coupling constants are in good agreement with those expected from the angles in a Dreiding model. The corresponding, not epoxidized lactone inunolide, without established stereochemistry, is reported from *I. racemosa* [24]. We therefore propose the name 4β ,5 α -epoxy-4,5-H-cis-inunolide for 20. The 1β ,10 α -epoxide we also have isolated. The structure follows again from the PMR data (see Table 4) and decoupling experiments.

21 is a lactone derived from the widely distributed hydrocarbon germacrene D. Again the PMR signals and the results of decoupling experiments are in good agreement with the proposed structure. Comparison of the couplings of 7- and 8-H with those of 20 indicates a trans-lactone (see Table 4).

Structures of 23 and 25 again follow from the PMR spectra. As the spectrum of 23 was not well resolved, the substance was transformed to the acetate 24. Addition

of Eu(fod), was again helpful in solving some of the stereochemical problems. Inspection of a Dreidingmodel shows that the proposed configurations agree with the observed coupling constants (see Table 5). In the case of 25 good results were only obtained in deuteriobenzene by using different amounts of shift-reagent at elevated temperature. As shown in Table 5 all signals then can be assigned. The relative shifts of the signals of 6α - and 6β -H are in better agreement with β - than with α-orientation of the epoxide. The same argument leads to 4β -orientation of the acetoxy group in 24. Similarly trans-annelation of the lactone rings agrees much better with the observed coupling constants than cis-annelation. Furthermore comparison of the PMR data of 23 with those of inuviscolide [6] shows that the configuration at C-4 is different, as in the latter case the signal of 5α -H is shifted downfield due to the cis-position of the 4-OH group. A lactone with the same structure, but without configurational assignment, has been isolated from Helenium species [7]. The few data given however, do not agree with those of 23. Therefore the stereochemistry must be different; 23 is 4-epi-isoinuviscolide.

The only pseudoguaianolide isolated from 4 *Inula* species was 26. Its structure and configuration follow from the PMR data (see Table 6) and from the result of

Table 1. PMR data of 1, 3, 6, 8 and 9 (270 MHz, CDCl₃, δ-values, TMS as external standard)

	1	3	6	8	9	+ Eu(fod)
1α-Η 1β-Η	ddd 1.25 ddd 1.61	m 1.26 d(br) 1.86	m 1.25 d(br) 1.91	d 2.28 dd 2.36	dd 3.30	dd 4.86
2α-Η 2 <i>β</i> -Η	} m 1.57	dddd 4.91	 dddd 4.19		m 1.58	d(br) 2.57 dddd 2.8
3α-H 3β-H	ddq 2.01 dddd 2.35	ddd 2.74 m 2.25		dd 2.6 ddd 2. 2 7	m 1.83 m 2.31	
3ρ-11 4α-Η		m 2.23	dug 2.65	ddq(br) 2.93	m 2.43	m 2.7
5α-H	$d(br) \ 0.85$	d(br) 2.06			with season in	
6α-H	ddd 1.75	ddd 1.94	15.25	1552	1.524	155
6β-H	ddd 1.34	ddd 1.40	d 5.25	d 5.52	d 5.24	d 5.5
7α-H	ddd(br) 2.98	ddddd 3.00	dddd 3.63	dddd 3.76	dddd 3.57	dddd 3.77
8α-H	ddd 4 51	ddd 4.51	ddd 4.82	ddd 4.88	ddd 4.86	ddd 5.09
9α-H	dd 1.5	dd 1.54	m 1.9	dd 1.79	dd 1.56	dd 2.24
9β-Η	dd 2.21	dd 2.27	dd 2.19	dd 2.24	dd 2.59	dd 3.61
13-H	d 6.14	d 6.16	d 6.21	d 6.28	d 6.22	d 6.51
13'-H	d 5.59	d 5.61	d 5.66	d 5.71	d 5.64	d 5.77
14-H	s 0.84	s 0.92	s 1.29	s 1.16	s 1.19	s 1.99
15-H	ddd 4.78	s(br) 4.94) , , , ,] ,,,,)	1,140
15'-H	ddd 4.45	s(br) 4.61	d 1.17	d 1.18	d 1.09	d 1.40
OAc		s 2.03	, <u></u>	,	, <u> </u>	´

 $J(\text{Hz}): 1: 1\alpha, 1\beta = 1\alpha, 2\beta = 12; 1\alpha, 2\alpha = 4; 1\beta, 2\alpha = 1\beta, 2\beta = 3; 2\alpha, 3\alpha = 6; 2\beta, 3\alpha = 12; 3\alpha, 3\beta = 13; 3\alpha, 15 = 1.5; 2\alpha, 3\beta = 2; 2\beta, 3\beta = 2; 3\beta, 15 = 1.5; 5\alpha, 6\beta = 12; 5\alpha, 6\beta = 2.5; 6\alpha, 6\beta = 14; 6\alpha, 7\alpha = 6.5; 6\beta, 7\alpha = 11; 7\alpha, 8\alpha = 5; 7\alpha, 13 = 1; 8\alpha, 9\alpha = 5; 8\alpha, 9\beta = 2; 3: 1\alpha, 2\beta = 13; 1\alpha, 2\beta = 12; 1\beta, 2\beta = 5; 2\beta, 3\alpha = 12; 2\beta, 3\beta = 5; 5\alpha, 6\beta = 12; 6\alpha, 7\alpha = 6; 6\beta, 7\alpha = 12; 7\alpha, 8\alpha = 5; 7\alpha, 13 = 1; 8\alpha, 9\alpha = 5; 8\alpha, 9\beta = 2; 9\alpha, 9\beta = 16; 6: 1\alpha, 2\beta = 10; 1\beta, 2\beta = 3; 2\beta, 3\alpha = 10: 2\beta, 3\beta = 3; 3\alpha, 4\alpha = 3\beta, 4\alpha = 4\alpha, 15 = 7; 6, 7\alpha = 4: 7\alpha, 8\alpha = 5; 7\alpha, 13 = 2; 8\alpha, 9\alpha = 4; 8\alpha, 9\beta = 2, 5, 9\alpha, 9\beta = 15; 8: 1\alpha, 1\beta = 15; 1\beta, 3\beta = 1; 3\alpha, 3\beta = 16; 4\alpha, 6 = 1; 6, 7\alpha = 5, 7\alpha, 8\alpha = 7; 7\alpha, 13 = 2.2; 7\alpha, 13' = 1.8: 8\alpha, 9\alpha = 2.5; 8\alpha, 9\beta = 3.5; 9\alpha, 9\beta = 16; 9: 1\alpha, 2\alpha = 4; 1\alpha, 2\beta = 2\alpha, 2\beta = 2\beta, 3\alpha = 12; 2\beta, 3\beta = 4; 4\alpha, 15 = 7; 6, 7 = 4; 7.8 = 6.5; 7.13 = 1.7; 8.9\alpha = 3.$

Table 2. PMR data of 10, 11 and 12 (CDCl₃)

	10	11	12	+ $\operatorname{Eu(fod)}_3$
1α-H 1β-H		dd 3.63	dd 3.7	1.13
2α-H (m 1.45-1.75) 1.74	dddq 2.4	0.65
2β-Н)			dddq 2.01	0.8
3α-H 3β-H	} m 1.97	} m 2	s(br) 5.34	0.21
5α-H	, –	, –	d(br) 2.17	0.36
	dd 2.81	dd 3	ddd 2.25	0.2
6 β -H	m 1.97	dd(br) 1.96	ddd 1.42	0.29
7α-H	ddddd 3.05	dddddd 2.25	ddddd 2.42	0.32
3-H	ddd 4.50	ddd 4.09	ddd 4.05	0.4
9α-Η) ,,,,,,	dd 1.46	dd 1.42	0.61
9β-Н	d 1.74	dd 2.58	dd 2.51	0.9
13-H	d 6.24	d 6.11	d 6.09	0.39
13'-H	d 5.6	d 5.45	d 5.45	0.17
14-H	s 1.08	s 1.09	s 0.89	0.53
15-H	s(br) 1.66	s(br) 1.67	s(br) 1.69	0.15

J(Hz): $10:6\alpha,6\beta=13.5; 6\alpha,7\alpha=7.5; 6\beta,7\alpha=10; 7\alpha,8\alpha=7.5; 7\alpha,13=2.5; 8\alpha,9=7.5; 11: 1\alpha,2\alpha=6; 1\alpha,2\beta=10; 6\alpha,6\beta=13.5; 6\alpha,7\alpha=3; 6\beta,7\alpha=12; 7\alpha,8\beta=11; 7\alpha,13=3; 8\beta,9\alpha=11; 8\beta,9\beta=3.5; 12: 1,2\alpha=7; 1.2\beta=10; 2\alpha,2\beta=12; 2\alpha,3=2\beta,3=2.15\sim2; 3.15=2; 5\alpha,6\alpha=3; 5\alpha,6\beta=12.5; 6\alpha,6\beta=12.5; 6\alpha,7\alpha=3; 6\beta,7\alpha=12.5; 7\alpha,13=3; 7\alpha,8\beta=12.5; 8\beta,9\alpha=12.5; 8\beta,9\beta=4.$

1-H

 $2\alpha - H$

2β-H $3\alpha-H$

 3β -H

5-H

6α-H

6β-H

7α-H

8-H

9β-H

9α-H

13-H

13'-H

14-H

15-H

+ Eu(fod),

d(br) 3.58

m 2

dd(br) 5.48

d(br) 2.75

ddddd 3.02

dd(br) 4.66

m 2.45

m 2.45

dd 3.13

dd 2.02

d 6.62

d 5.72

s 1.71

dd 1.87

Table 3. PMR data of 14-18 (CDCl.)

Table	4. PMR	data	of 20 ,	21	and	22
20		21			22	

dd(br) 5.36

m2.16

d(br) 2.82

d(br) 6.07

dddd 3.6

ddd 4

dd 5.33

dd 2.64

d 6.14

d 5.48

d(br) 2.82

s(br) 1.56

(s(br) 4.96

s(br) 4.8

dd 2.67

m 1.45

m 2.25

dd(br) 5.2

d(br) 2.51

dd(br) 4.17

d(br) 2.73

m 2.25

m2.71

m 1.45

d 6.29

d 5.62

s 1.28

dd 1.69

dd(br) 5.27

m 2.39

m2.2

ddd 1.25

d(br) 2.65

ddd 1.45

ddd 4.1

d(br) 2.79

s(br) 1.78

m 2.35

m 2.85

m 2.34

d 6.34

d 5.65

s 1.17

	14	Δ†	16‡	
1-H	dd(br) 5.07	0.30	dd(br) 5.19	
2-H)) ´	
3-H	m 2.3		m 2.32	
5-H	d(br) 4.61	0.26	d(br) 4.69	
6β-Н	dd 4.54	0.33	dd 4.57	
7α-H	ddddd 2.65	0.56	ddddd 2.75	
8α-H)			
8β-Н	}m 1.95		m 2.05	
9α-Η	dd 4.19	0.91	dd(br) 5.22	
13-H	d 6.27	0.30	d 6.31	
13'-H	d 5.57	0.18	d 5.56	
14-H	s(br) 1.44	0.42	s(br) 1.46	
15-H	s(br) 1.7	0.16	s(br) 1.74	
OCOR			tq 2.31	
			d 1.15	
			t 0.90	

[†] Δ-Values after addition of 0.1 equivalents of Eu(fod)₃.

J (Hz): 20: 1, $2\alpha = 3$; 1, $2\beta = 12$; 2α , $3\alpha = 7$; 2β , $3\alpha = 13$; 3α , $3\beta = 13$; 5.6 = 10; $6\alpha,6\beta = 15$; $6\beta,7\alpha = 10$; $7\alpha,8\alpha = 7$; 8α , $9\beta = 11$; 8α , $9\alpha = 1.5$; 9α , $9\beta = 13$; 7, 13 = 3.5; 7, 13 = 3.0; **21**: 1, $2\alpha = 2$; 1, $2\beta = 10$; 3α , $3\beta = 13$; 5, 6 = 16: 6, $7\alpha = 9$;

 7α , $8\beta = 9.5$; 8β , $9\alpha = 10$; 8β , $9\beta = 3$; 7, 13 = 3.5; 7, 13 = 3.0; **22**: 1, $2\alpha = 10$; 1, $2\beta = 2.3$; 5, $6\alpha = 10$; 5, $6\beta = 4$; 6α , $6\beta = 14$; 6α , $7\alpha = 3.5$; 6β , $7\alpha = 10$; 7α , $8\alpha = 7$; 7α , 13 = 3.3; 8α , $9\beta =$ 10.5; 9α , $9\beta = 13.5$.

‡ The values of the other esters are very similar. J(Hz): 14: 1,2 α = 11; 1, 2 β = 2.5; 5, 6 = 10; 6, 7α = 9.5; 7α , $8\alpha \sim 2$; 7α , $8\beta = 9$; 7α , 13 = 3.5; **15**: OCOR: q = 2.35(2) t = 1.14(3)(J=7) 17: d 2.37(2) m 1.5(1) d 1.00(6) (J=7); 18: qq 2.54(2), d

1.17(3), d 1.16(3) (J = 7).

Table 5. PMR data of 23, 24 and 25

	23	24	+ Eu(fod) ₃	$25 \left(\mathrm{C_6D_6} \right)$	+ Eu(fod) ₃ (80°
1-H)	m 2.07	m 2.11	d(br) 2.15	d(br) 4.51
2α-H	1))	m 1.7	dd(br) 3.36
2β-H	m 1.5-1.9	m 1.77	m 1.94	dd 1.72	dddd 3.52
3α-H	j	m 2.07	m 2.69	dd 0.85	dd 2.2
3β-H	m 2.5	dd(br) 2.44	m 2.3	m 1.34	ddd 2.81
5-H	m 1.95	m 2.07	$d(br) \ 2.43$		_
6α-Η	m 2.5	ddd 2.75	ddd 3.33	d(br) 1.77	d(br) 5.29
6β-H	ddd 1.43	ddd 1.41	ddd 1.73	m 1.34	dd(br) 3.92
7α-H	ddddd 2.54	ddddd 2.56	t(br) 2.81	ddddd 2.36	dddddd 5.21
8 <i>β</i> -H	ddq 4.72	ddq 4.72	d(br) 4.95	ddd 3.41	ddd 6.23
9α-Η]))	ddd 1.31	ddd 2.96
9β-Н	\ddq 5.84	\ddq 5.87	s(br) 6	ddd 1.89	ddd 3.17
10α-H				m 1.43	m 2.7
13-H	d 6.19	d 6.2	d 6.45	d 6.07	d 9.31
13'-H	d 5.50	d 5.55	d 5.68	d 4.81	d 6.26
14-H	s(br) 1.82	s(br) 1.79	s(br) 1.89	d 0.41	d 1.75
15-H	s 1.23	s 1.44	s 2.04	s 1.1	s 2.96
OAc	_	s 1.99	s 2.75		

J (Hz): **24**: 5α , $6\alpha = 6\alpha$, $7\alpha = 3$; 5α , $6\beta = 6\alpha$, $6\beta = 13$; 6β , $7\alpha = 11$; 7α , $8\beta = 9.5$; 7α , 13 = 3.5; 7α , 13' = 3; 8β , 9 = 1.5; 8β , 14 = 1.5; 9, 14 = 1.5; **25**: 1, $2\alpha = 8$; 2α , $2\beta = 13$; 2α , $3\alpha = 9$; 2β , $3\beta = 8$; 2α , $3\beta = 10$; 3α , $3\beta = 13$; 6α , $6\beta = 16$; 6α , $7\alpha = 3$; 6β , $7\alpha = 10$; 7α , $8\beta = 10$; 7α , 13' = 10; 13; 14; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 15; 3.5; 7α , 13 = 3; 8β , $9\alpha = 12$; 8β , $9\beta = 3.5$; 9α , $9\beta = 12$; 9α , $10\alpha = 3$; 9β , $10\alpha = 4$; 10α , 14 = 7.

the chromic acid/pyridine oxidation, which leads to a ketone whose physical properties are in good agreement with those reported for confertin 27 [8].

In addition to some known xantholides 3 other

lactones were isolated. 29 is probably identical with tomentosin, isolated from Parthenium tomentosum [9]. 29 is also present in Inula viscosa Ait. [6]. We now also have isolated the 8-epimer 32, which may be identical

Table 6. PMR data of 26

		+ Eu(fod) ₃
3α-H	m 2.15	m 4.73
3β -H	m 1.45	m 3.61
4β-H	dd 3.76	dd 9.38
6α-H	d(br) 1.62	d(br) 6.18
6β-H	dd 1.87	dd 4.28
7α-H	ddddd 3.15	m 5.22
8α-H	ddd 4.7	ddd 6.39
9α-H	d(br) 1.86	$d(br) \ 3.25$
9β-H	ddd 2.15	ddd 3.43
10-H	m 1.45	m 3
13-H	d 6.27	d 8.4
13'-H	d 5.6	d 6.28
14-H	d 1.03	d 1.88
15-H	s 0.92	s 3.54

J (Hz): 3α, 4β = 8; 3β, 4β = 8; 6α, 6β = 13; 6β, 7α = 13; 7α, 8α = 8; 8α, 9α = 3; 8α, 9β = 12; 9α, 9β = 13; 9β, 10α = 13; 10α, 14 = 7

with xanthinosin, found in a Xanthium species [22]. Data, however, are not given. Careful PMR studies lead to the conclusion that most probably tomentosin has a cisfused lactone ring and that 32 is the trans-isomer. As already mentioned previously the chemical shift of the 7α -H seems to be characteristic, while $J_{7.8}$ is not useful in making a distinction between cis- and trans-lactone ring fusion. Also the CD-curves of 29 and 32 are in agreement with the stereochemistry assigned.

We have also isolated the corresponding alcohol 28, which on oxidation affords 29. Together with carabrone (31) [10] we have also found the corresponding alcohol 30, which could be oxidized to a ketone identical with

31. The configuration at C-4 in alcohols 28 and 30 has not been established. PMR data of 28-30 and 32 are given in Table 7.

The compounds in scheme 2 are all known except for 40 whose structure clearly follows from the spectroscopic data (see Experimental).

In summary, it turns out that again two groups of *Inula* species can be recognized, one containing sesquiterpene lactones and one containing mainly thymol derivatives.

I. helenium L. which has been previously investigated [10] afforded a considerable number of lactones with different carbon skeletons. I. royleana DC, from which so far only diterpenes have been reported [11], also contained several different types of lactones. There were some differences between the constituents of plants which have grown for 2 and 5 years. I. salicina L. afforded a series of thymol derivatives as already reported earlier [12], together with small amounts of 1 and 5. I. bifrons L. yielded only thymol derivatives.

From the chemotaxonomical point of view the isolation of very different types of sesquiterpene lactones may be of interest. However, while the main constituents are eudesmanolides, the other types are only present in relatively small amounts. Clearly further investigations are necessary to get a clearer picture about this large genus.

EXPERIMENTAL

IR: CCl_4 or $CHCl_3$ solns; PMR: 270 MHz δ -values, TMS as internal standard; optical rotation. $CHCl_3$ solns. The plant material was grown from seed and fr. material was extracted with Et_2O petrol 1:2. The extracts were first separated by column chromatography (Si gel grade II activity) and further by

Table 7. PMR data of 28, 29, 30 and 32

	28	32	+ Eu(fod) ₃	29	30
1-H			_		m 0.38
2-H)		dd(br) 3.44) 242	
2'-H	t 2.25	1	dd(br) 3.33	m 2.43	
	,	m 2 53			
3-H	t 2.53		dd(br) 3.88	m 1 54	`
3'-H	∫t 2.33		dd(br) 3.78	}m 1 54	\dt 1.54
	•	J		,	}
4-H	_			tq 3.8	tq 3.48
5-H	dd(br) 5.45	dd(br) 5.53	dd(br) 6.09	dd(br) 5.5	m 0.47
6α-H	2.30	m 2.3	dd(br) 2.42	ddd 1.92	٠ —
6β-H	m 2.38	dd(br) 2.02	ddd 2.88	ddd 2.2	dd 2.38
7α - H	ddddd 3.32	m 2.5	ddddd 3.13	ddddd 3 34	ddddd 3.15
8-H	ddd 4.65	ddd 4.24	ddd 4.85	ddd 4.67	ddd 4.8
9α-H	ddd 1.9	ddd 1.77	ddd 2.22		m 1.2
9β-H	ddd 2.02	ddd 2.28	ddd 2.59	m 2	dd 2.32
10-H	m 2.38	m 2.3	ddq 2.99		
13-H	d 6.26	d 6 16	đ 7.08	d 6.27	d 6.25
13'-H	d 5.53	d 5.43	d 5.85	d 5.53	d 5.55
14-H	d 1.14	d 1.14	d 1.39	d 1.14	s 1.08
15-H	s 2.16	s 2.16	s 3.4	d 1.2	d 1.21

J (Hz): 32: 2, 2' = 16; 2, 3 = 7; 3, 3' = 15; 5, $6\alpha = 3$; 5, $6\beta = 8.5$; 6α , $6\beta = 15.5$; 6α , $7\alpha = 8.5$; 6β , $7\alpha = 10$; 7α , $8\beta = 11$; 8β , $9\alpha = 12.5$; 8β , $9\beta = 3$; 9α , 10 = 4; 9β , 10 = 3; P (Hz): 29: 3, 4 = 4, 15 = 7; 5, $6\alpha = 6$; 5, $6\beta = 9$; 6α , $7\alpha = 4$; 6β , $7\alpha = 12$; 7α , 13 = 3; 7α , $8\alpha = 8$; 10, 15 = 7; 30: 3, 4 = 4, 15 = 6; 6α , $6\beta = 14$; 6α , $7\alpha = 6\beta$, $7\alpha = 7$; 7α , $8\alpha = 11.5$; 7, 13 = 3; 7, 13' = 2.5; 8α , $9\alpha = 6$; 8α , $9\beta = 9$; 9α , $9\beta = 14$

Table 8. Investigated species and isolated constituents

	kg	Roots	kg	Aerial parts
Inula helenium L. (seeds Bot. Garden Budapest, voucher 77/952, 2 years grown)	3	6 g 1, 2 g 5, 550 mg 10*, 100 mg 21*, 20 mg 33, 200 mg 35, 180 mg 36, 15 mg 45	1.5	20 mg 1, 10 mg 4, 30 mg 5, 15 mg 8†, 4 mg 9‡, 60 mg 12‡, 30 mg 13, 30 mg 14‡, 30 mg 16†, 3 mg 17†, 6 mg 18†, 50 mg 20†, 25 mg 22†, 10 mg 23‡, 40 mg 25†, 2 mg 26‡, 40 mg 29, 60 mg 31‡, 25 mg 32†
Inula royleana DC (seeds Bot. Garden Vacratot) (a) 5 years grown voucher 77/960	1.5	4 g 1, 3 g 5 5 mg 33, 100 mg 36	2	70 mg 1, 40 mg 2, 10 mg 3†, 75 mg 6, 11 mg 10, 50 mg 11‡, 3 mg 12, 17 mg 14, 20 mg 15†, 20 mg 16†, 8 mg 20, 120 mg 25, 7 mg 28‡, 3 mg 29, 69 mg 30†, 4 mg 31
(b) 2 years grown voucher 77/991	2	7 g 1, 5 g 5, 2 mg 33, 20 mg 36	2	35 mg 1, 25 mg 5, 10 mg 6, 2 mg 8, 3 mg 20, 28 mg 23, 8 mg 30, 8 mg 31
Inula salicina L. (seeds Bot. Garden Triest, voucher 77/990) (1 year grown)	0.05	0.1 mg 33, 10 mg 34, 2 mg 37, 50 mg 38, 20 mg 39, 10 mg 41, 15 mg 42, 2 mg 43, 4 mg 44, 35 mg 45, 60 mg 46	0.2	20 mg 1, 50 mg 5, 13 mg 38, 8 mg 39, 7 mg 41, 5 mg 48
Inula bifrons L. (seeds, Bot. Garden Vacratot, voucher 77/982 (5 years grown)	0.4	12 mg 33, 15 mg 35, 9 mg 38, 20 mg 40 (ether/petrol 3:1) 18 mg 43, 120 mg 45, 24 mg 47	3	35 mg 34, 15 mg 49, 200 mg linolenic acid

^{*} Ag-NO₃ coated plates, ether/petrol 2:1; † ether/petrol 1:1; and CH₂Cl₂/C₆H₆ 2:1; ‡ ether/petrol 2:1, several times.

TLC (Si gel, GF 254), which was repeated several times. In some cases only AgNO₃-coated plates were successful. Known compounds were identified by comparison of IR- and PMRspectra with those of authentic material. The results are given in Table 8.

Ivalin-acetate (3). Colourless oil, IR: OAc 1730, 1260, methylene lactone 1770, 1660 cm^{-1} . MS: M^+ m/e -; $(M^+\text{-AcOH})$ 230.131 (48%) ($C_{15}H_{18}O_2$); 230-'Me 215(8); $C_9H_{11}^+$ 119(100). 3 was identical with the acetate of ivalin (IR, PMR, TLC).

2α-Hydroxy alantolactone (6). Colourless oil, IR: OH 3620; methylene lactone 1770, 1660 cm⁻¹. MS: M⁺ m/e 248.141 (5%) $(C_{15}H_{20}O_3)$; (M^+-H_2O) 230(37); 230-'Me 215(12); $C_3H_7^+$ 43(100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{+60.7} \frac{578}{+62.2} \frac{546}{+71.8} \frac{436}{+134.5} (c = 1.0).$$

20 mg 6 in 2 ml CH₂Cl₂ were stirred at room temp with 100 mg CrO₃/pyridine complex. After TLC of the reaction product 13 mg 8 were obtained, identical with the natural product. 10 mg 6 on heating with 1 ml Ac₂O at 70° for 30 min afforded 8 mg 7, colourless oil, IR: OAc 1750, 1245, lactone 1765 cm⁻¹.

2-Oxo-alantolactone (8). Colourless crystals from Et₂O, mp 152-153°. IR: C=O 1715; methylene lactone 1765 cm⁻¹ MS: M⁺ m/e 246.157(100%) (C₁₅H₁₈O₃); (M⁺-Me) 231(56); (M^+-H_2O) 228(2); $(M^+-H_2C=C=O)$ 204(26); (228-Me) 213(8); (204-CO) 176(82).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{+301.3} \frac{578}{+316.4} \frac{546}{+366.5} \frac{436 \,\mathrm{nm}}{+700.6^{\circ}} (c = 1.3).$$

1β-Hydroxy alantolactone (9). Colourless oil, IR: OH 3620; methylene lactone 1790, 1675 cm⁻¹. MS: M+ m/e 248.141(29%) $(C_{15}H_{20}O_3)$; (M^+-H_2O) 230(36); (230-Me) 215(37); $C_3H_7^+$

1-Desoxy-8-epi-ivangustin (10). Colourless oil, IR: methylene

lactone 1780, 1670 cm⁻¹. MS: M⁺
$$m/e$$
 232.146 (21%) (C₁₅H₂₀O₂)

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{+35.2} \frac{578}{+36.4} \frac{546}{+40.0} \frac{436}{+59.6^{\circ}} (c = 0.5)$$

8-Epi-ivangustin (11). Colourless oil, IR: OH 3620; methylene lactone 1780, 1670 cm⁻¹. MS: M⁺ m/e 248.141 (72%) (C₁₅H₂₀O₃); (M+-H,O) 230(67); (230-Me) 215(100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{-25.4} \frac{578}{-27.4} \frac{546}{-31.6} \frac{436}{-56.9} (c = 3.7).$$

8-Epi-isovangustin (12). Colourless crystals, mp 135.5° (Et₂O/ petrol), IR: OH 3620; methylene lactone 1775 cm⁻¹. MS: M+ m/e 248.141 (14%) (C₁₅H₂₀O₃); (M⁺-H₂O) 230(11); (230-Me) 215(8); C₆H₉⁺ 81(100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{-50.4} \frac{578}{-52.5} \frac{546}{-59.8} \frac{436 \text{ nm}}{-102.5^{\circ}} (c = 0.48).$$

9β-Hydroxy costunolide (14). Colourless oil, IR: OH 3610; methylene lactone 1760, 1670 cm⁻¹. MS: $M^+ m/e$ 248.141 (25%) $(C_{15}H_{20}O_3)$; $(M^+$ -'Me) 233(5); $(M^+$ -H₂O) 230(26); (230-'Me) 215(15); $C_9H_{11}^+$ 119(100).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{+38.6} \frac{578}{+42.7} \frac{546}{+61.5^{\circ}} (c = 1.0).$$

9β-Propionyloxy costunolide (15). Colourless oil, IR: CO₂R 1725; methylene lactone 1770, 1665 cm⁻¹. MS: M⁺ m/e 304.167 $(5\%)(C_{18}H_{24}O_4);(M^+-C_2H_5CO_2H)230(51);(230-Me)215(23);$ $C_2H_5CO^+$ 57(100).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{+54.3} \frac{578}{+56.4} \frac{546}{+64.3} \frac{436 \text{ nm}}{+98.8^{\circ}} (c = 0.6).$$

 9β -(2-Methyl butyryloxy)-costunolide (16). Colourless oil, IR: CO₂R 1725; methylene lactone 1770, 1665 cm⁻¹. MS: M⁺ m/e $332\overline{1}99 (20\%) (C_{20}H_{28}O_4); (M^+-C_4H_9CO_2H) 230(100); (230-4)$ 'Me) 215(41).

9B-Isovaleryloxy costunolide (17). Colourless oil, not free

from 18, IR: CO_2R 1725; methylene lactone 1770, 1665 cm $^{-1}$. MS. M $^+$ m/e 332.199 (17%) ($C_{20}H_{28}O_4$).

9β-Isobutyryloxy costunolide (18). Colourless crystals, mp 154° (Et₂O/petrol), IR: CO₂R 1725; methylene lactone 1770, 1660 cm⁻¹. MS: M⁺ m/e 318.183 (6%) (C₁₉H₂₆O₄); (M⁺-C₃H₇CO₂H) 230(49); (230-Me) 215(31); C₃H₇CO⁺ 71(100).

$$\label{eq:alpha_scale} \left[\alpha\right]_{24}^{\lambda} = \frac{589}{+104.8} \, \frac{578}{+110.6} \, \frac{546}{+128.3} \, \frac{436 \, \mathrm{nm}}{+245.2^{\circ}} (c = 0.48).$$

4 β .5 α -Epoxy-4.5-cis-inunolide (20). Colourless crystals from Et₂O, mp 131°. IR: methylene lactone 1775, 1665 cm $^{-1}$. MS: M $^{+}$ m/e 248.141 (17%) (C₁₅H₂₀O₃); (M $^{+}$ -H₂O) 230(31); (230-Me) 215(12); (M $^{+}$ -C₃H₆O) 190(100).

$$\left[\alpha\right]_{24^{\circ}}^{\lambda} = \frac{589}{-14.4} \frac{578}{-15.2} \frac{546}{-17.5} \frac{436 \text{ nm}}{-18.4^{\circ}} (c = 1.0).$$

Germacrene-p-lactone (21). Colourless crystals, mp 142° (Et₂O/petrol). IR: methylene lactone 1775; C=C 1670, 1636, 995, 950, 895 cm $^{-1}$. MS: M* m/e 230.131 (18%) (C₁₅H₁₈O₂). (M*-'Me) 215(12); (M*-CO) 202(10); (M*-CHO) 201(16); C₇H₇+ 91(100).

$$\left[\alpha\right]_{24^{\circ}}^{\lambda} = \frac{589}{-200.9} \frac{578}{-213.2} \frac{546}{-255.9} \frac{436 \text{ nm}}{-590.6^{\circ}} (c = 1).$$

 1β ,10 α , Epoxy-1 10-H-cis-inunolide (22). Colourless crystals, mp 156° (Et₂O/petrol). IR: methylene lactone 1765 cm $^{-1}$. MS: M $^+$ m/e 248.141 (8%) (C₁₅H₂₀O₃); (M $^+$ -H₂O) 230(5); (230 - Me) 215 (4); (-C₃H $^+_7$) 43 (100).

$$\left[\alpha\right]_{24^{\circ}}^{\lambda} = \frac{589 \quad 578 \quad 546 \quad 436 \text{ nm}}{-20.0 \quad -21.5 \quad -25.5 \quad -48.0^{\circ}} \ (c = 0.2).$$

4-Epi-iso inuriscolide (23). Colourless crystals, mp 147°. IR: OH 3620: methylene lactone 1775 cm $^{-1}$. MS. M $^+$ m/e 248.141 (27%) (C $_{15}$ H $_{20}$ O $_{3}$); (M $^+$ -H $_{2}$ O) 230(9); (230°-Me) 215(8); C $_{3}$ H $_{43}$ (100). 20 mg 23 in 0.5 ml Ac $_{2}$ O and 20 mg 4-pyrrolidino-pyridine were heated for 30 min at 70°. After TLC (Et $_{2}$ O/petrol 3:2) 15 mg 24 were obtained, colourless oil, IR: OAc, methylene lactone 1770 cm $^{-1}$. MS: M $^+$ m/e 290.152 (6%) (C $_{17}$ H $_{22}$ O $_{4}$); (M $^+$ -AcOH) 230(70); (230-'Me) 215(21): H $_{3}$ CCO $^+$ 43(100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{-112.9} \frac{578}{-117.8} \frac{546}{-135.1} \frac{436 \text{ nm}}{-241.1^{\circ}} (c = 1.8).$$

 $^{4\alpha,5\alpha,Epoxy-10\alpha,14H-inuviscolide}$ (25) Colourless oil, 1R: methylene lactone 1775, 1660 cm $^{-1}$. MS: M $^+$ $^+$ $^+$ $^ ^-$ 248.141 (9 %) (C $_{15}H_{20}O_3$); (M $^+$ -'Me) 233(3); (M $^+$ -H $_2$ O) 230(26); (230-'Me) 215(13); (M $^+$ -C $_3H_6$ O) 190(100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{+39.7} \frac{578}{+42.0} \frac{546}{+48.0} \frac{436 \text{ nm}}{+84.4} (c = 1.4).$$

 $4\alpha\text{-H-}confertin\,(\textbf{26}).$ Colourless crystals, mp 147°. IR: OH 3605: methylene lactone 1760, 1660 cm $^{-1}.$ MS: M $^+$ m/e 250.157 (4 %) (C $_{15}\text{H}_{22}\text{O}_3$); (M $^+\text{-H}_2\text{O})$ 232(15); 108(100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{+83.3} \frac{578}{+88.8} \frac{546}{+101.3} \frac{436 \, nm}{+178.3^{\circ}} (c = 0.6).$$

6 mg **26** were oxidized with CrO₃/pyridine complex as above. After TLC (Et₂O/petrol 2:1) 3 mg **27** were obtained, identical data with those of confertin.

4H-Tomentosin (28) Colourless oil, IR: OH 3600; methylene lactone 1770, 1650 cm $^{-1}$. MS: M $^+$ m/e 250.15711 $^-$ IIC $_{11}$, H $_{22}$ O $_{3}$); (M $^+$ -H $_{2}$ O) 232(26); C $_{9}$ H $_{13}^+$ 121(100); C $_{7}$ H $_{9}$ =3(42)

$$[\alpha]_{24}^{\lambda} = \frac{589}{+37.0} \frac{578}{+38.8} \frac{546}{+43.7} \frac{436 \text{ nm}}{+68.6^{\circ}} (c = 0.9).$$

10 mg 28 were oxidized as above. After TLC 6 mg 29 were obtained, identical with authentic material.

4H-Carabrone (30). Colourless oil, IR: OH 3610; methylene lactone 1760, 1665 cm $^{-1}$. MS: M $^+$ m/e 250.157(1%)($C_{15}H_{22}O_3$); (M $^+$ -H $_2$ O) 232(2); C_5H_9 O 85(100).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{+19.4} \frac{578}{+20.4} \frac{546}{+25.5} \frac{436 \text{ nm}}{+56.0^{\circ}} (\epsilon = 1.0).$$

10 mg 30 on oxidation afford 6 mg 31, identical with authentic material

8-Epi-tomentosin (32) ('xanthinosin' [22]). Colourless oil, IR $^{\circ}$ C=O 1710; methylene lactone 1775 cm $^{-1}$. MS: M $^{+}$ m/e 248.141 (20%) ($C_{15}H_{20}O_{3}$): (M $^{+}$ -H $_{2}O$) 230(21); (230-'Mc) 215(7); (M $^{+}$ -CH $_{3}$ COCH $_{3}$) 190(53); $C_{3}H_{7}^{+}$ 43(100).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{-38.9} \frac{578}{-41.1} \frac{546}{-47.0} \frac{436 \text{ nm}}{-81.1^{\circ}} (c = 0.44).$$

2-Methoxy thymol isobutyrate (40) Colourless oil, IR: Ph-O-CO-R 1760; 1625, 1590 cm⁻¹. MS: M⁺ m/e 250.157 (7%) (C₁₅H₂₂O₃); (M⁺-Me₂C=C=O) 180(100); (180-Me) 165(61); C₃H₇CO⁺ 71(6); (71-CO) 43(25). PMR: s 6.72(3-H); s(br) 6.74(6-H); s 2.16(7-H); qq 2.97(8-H); d 1.33(9,10-H); qq 2.87 and d 1.20 (J = 7 Hz) (COCHMe₂); s 3.83 (OCH₃).

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